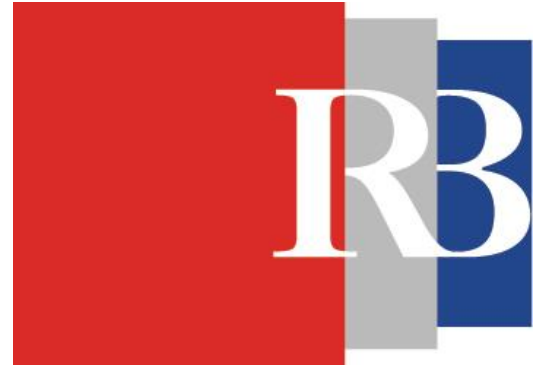


Engineered nanoparticles in the marine environment – the role of microalgal extracellular polymers



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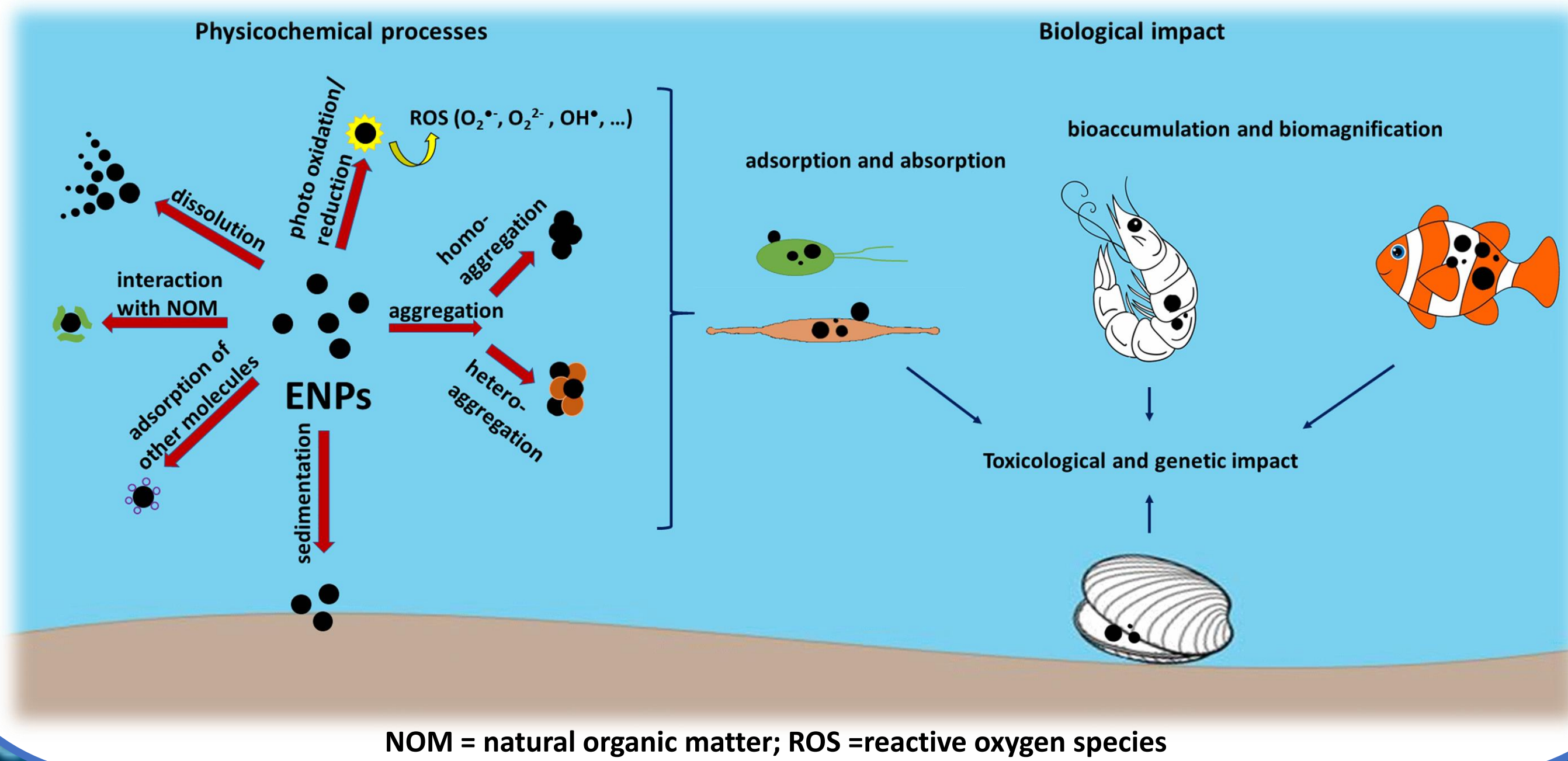


Division for Marine and Environmental Research

INTRODUCTION

Engineered nanoparticles (ENPs) constitute a major component of the global production of nanomaterials. Among them, silicon dioxide nanoparticles (SiO₂ NPs) have found extensive applications in a variety of industries (e.g. in cosmetics, drugs, toners, car tire manufacturing, etc.) [1,2]. The wide application of various ENPs causes their increasing presence in seawater.

Potential physicochemical processes and biological impacts of ENPs in seawater

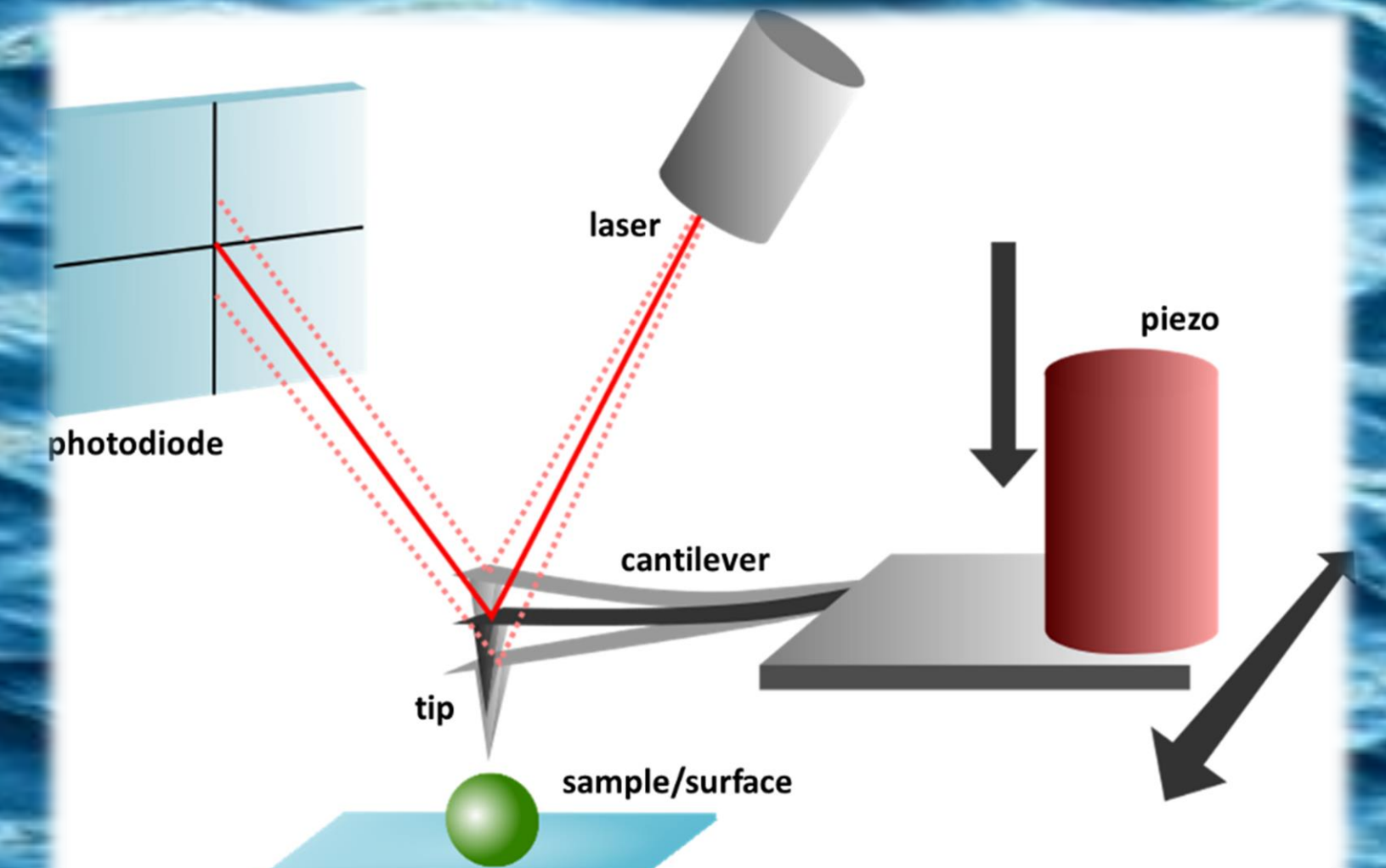


AIM

We studied the effect of microalgal EPS (extracellular organic substance), as the main component of marine natural organic matter, on the stability of silicon dioxide nanoparticles (SiO₂ NPs) in seawater.

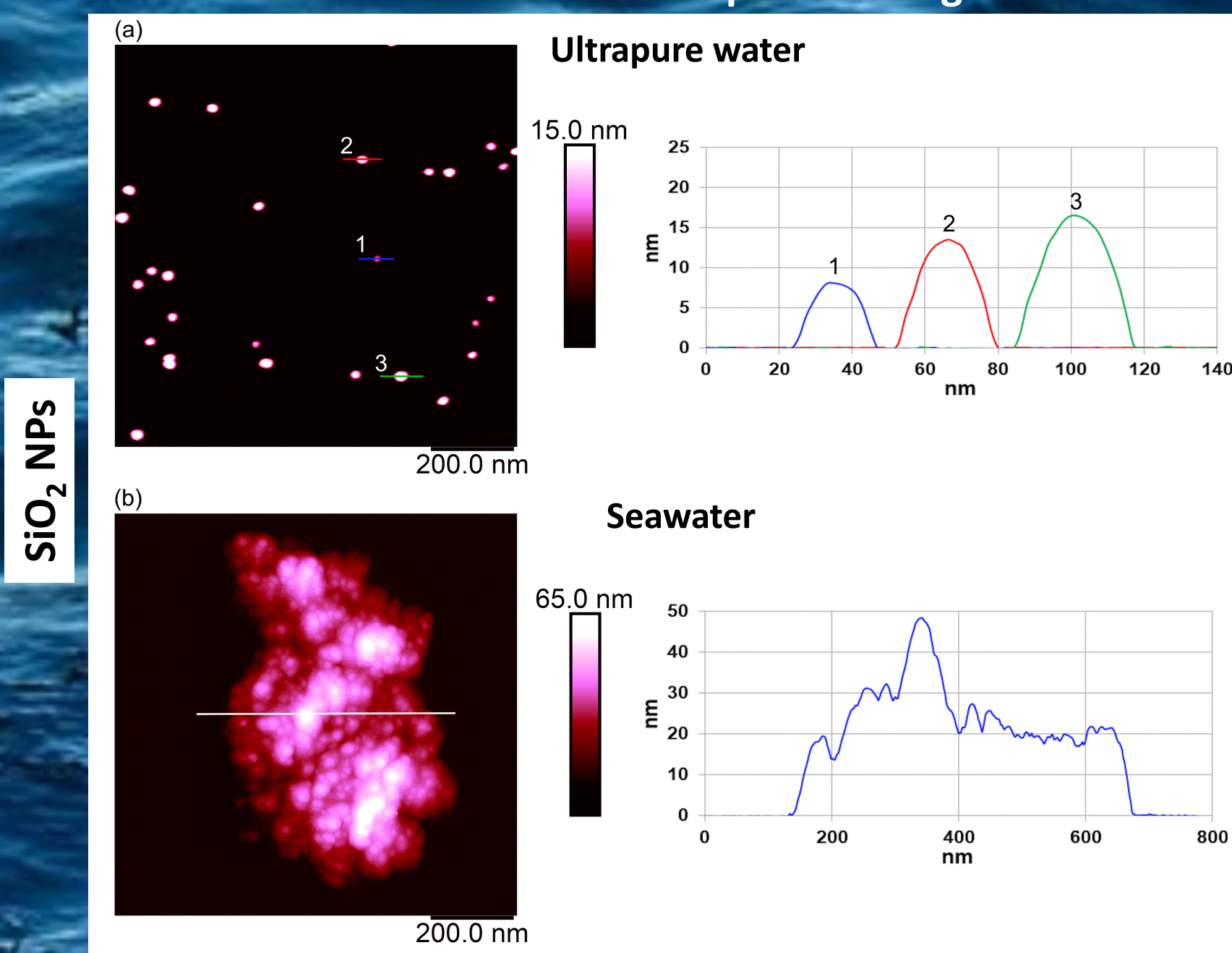
METHODOLOGY

Atomic force microscopy (AFM) Multimode AFM with Nanoscope IIIa controller (Bruker)

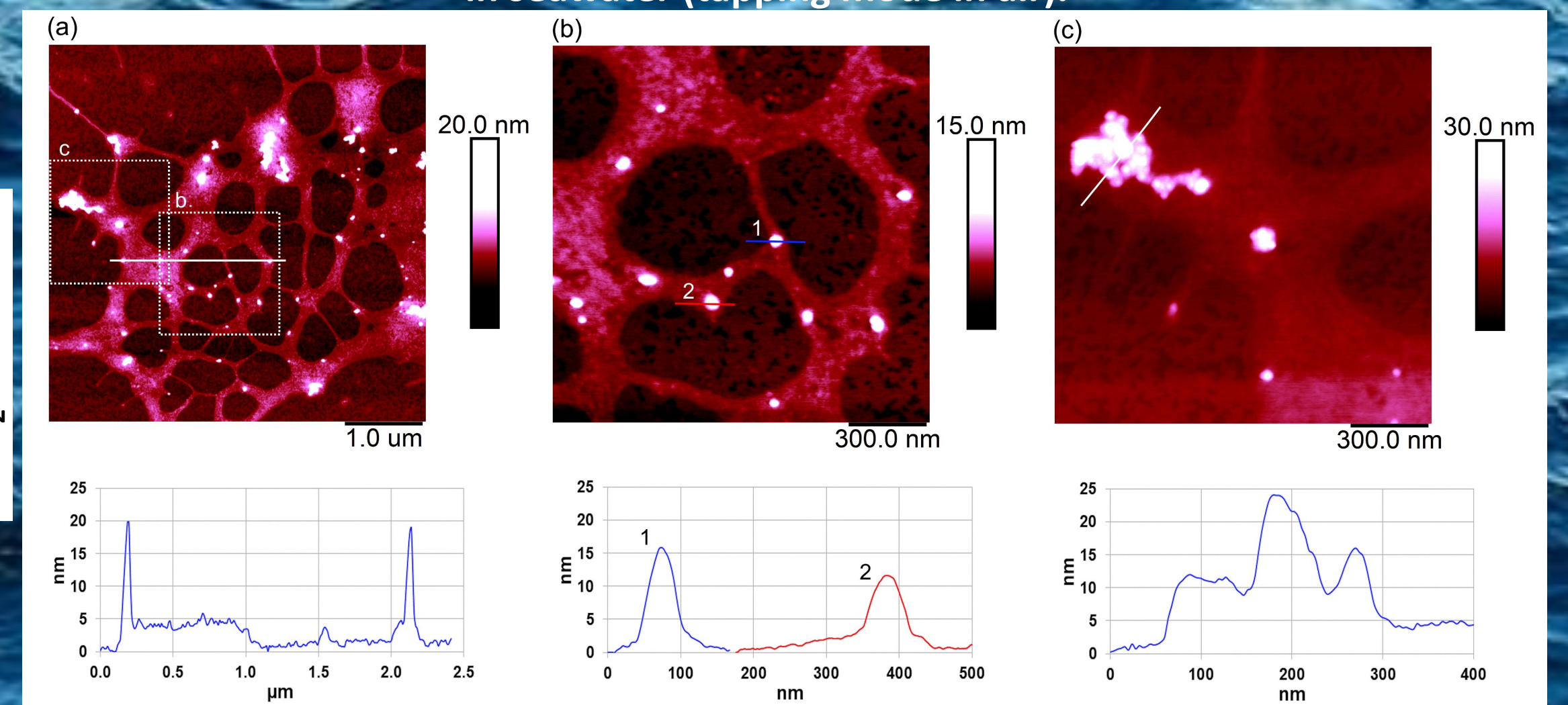


RESULTS

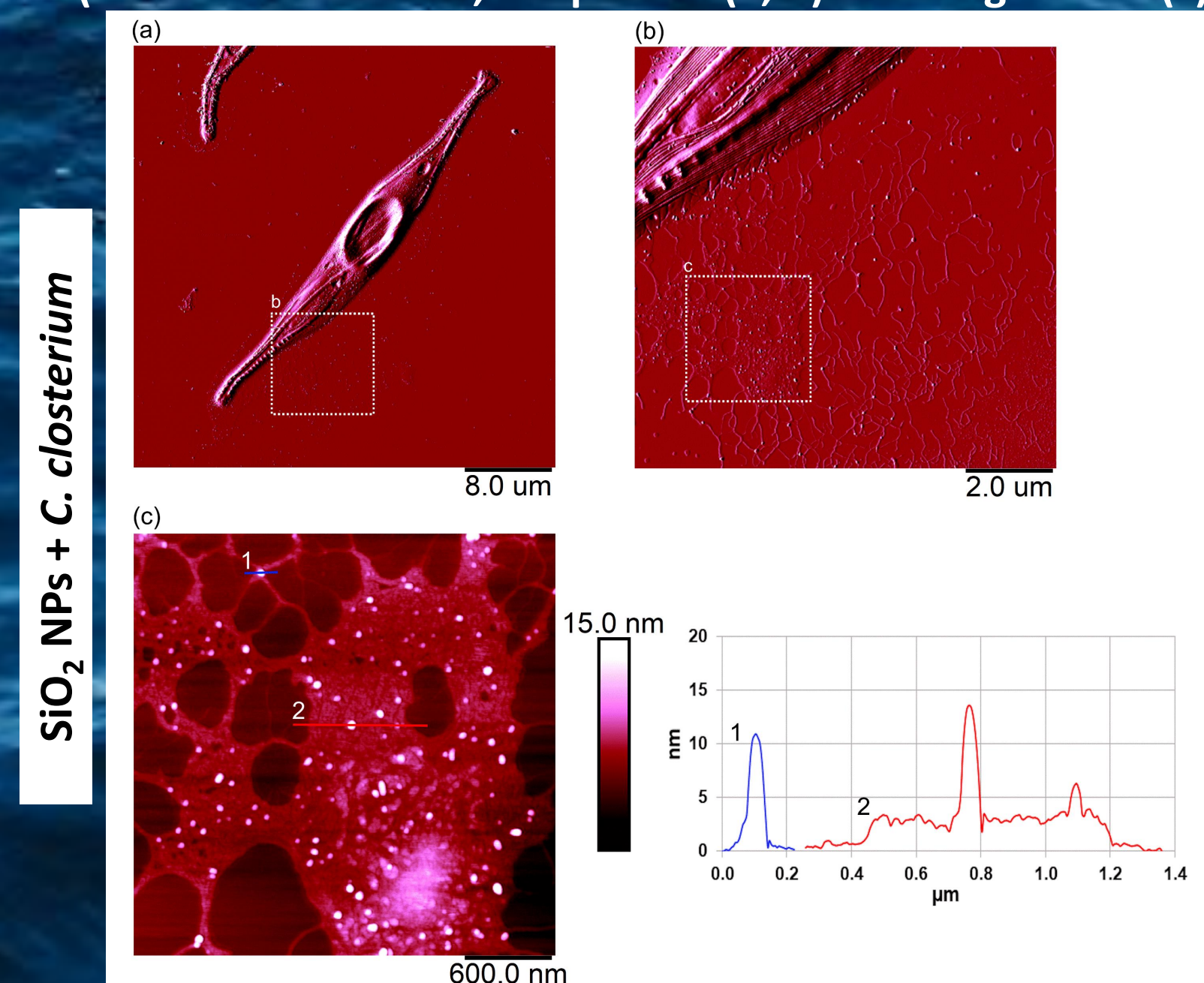
AFM images of SiO₂ NPs (10 μg mL⁻¹); tapping mode in air. Vertical profiles along indicated lines show particle heights.



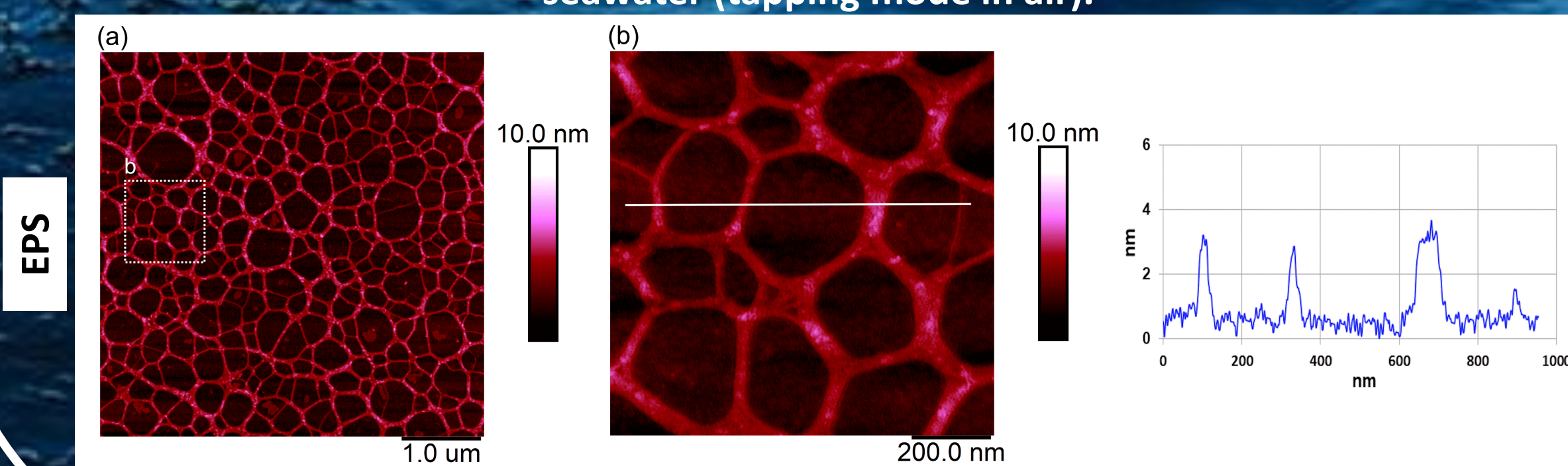
AFM images of interaction of EPS (200 μg mL⁻¹) with SiO₂ (10 μg mL⁻¹) in seawater (tapping mode in air).



AFM images of *C. closterium* cell exposed to SiO₂ NPs (10 μg mL⁻¹) in seawater (contact mode in air, amplitude (a, b) and height data (c)).



AFM images of EPS (200 μg mL⁻¹, isolated from *Cylindrotheca closterium* culture) in seawater (tapping mode in air).



CONCLUSIONS

- ❖ SiO₂ NPs are present mainly as individually dispersed nanoparticles in ultrapure water, while in filtered natural seawater as microscale aggregates
- ❖ SiO₂ NPs showed strong interaction with both, isolated EPS, and at the single-cell level, with EPS released around the cell of *Cylindrotheca closterium*
- ❖ EPS has potential to scavenge and stabilize SiO₂ NPs, thus prolonging their residence time in the water column what could have significant implications, especially during microalgal blooms when EPS production increases
- ❖ our results could also explain the persistence in the water column not only of SiO₂ NPs, but most likely also of other oxide types of engineered NPs (like TiO₂, CeO₂, Ag₂O, etc.)

REFERENCES

1. Napierska, D.; Thomassen, L.C.; Lison, D.; Martens, J.A.; Hoet, P.H. The Nanosilica Hazard: Another Variable Entity. *Part. Fibre Toxicol.* 2010, 7, 39.
2. Van Hoek, J. et al. Implications of the Use of Silica as Active Filler in Passenger Car Tire Compounds on Their Recycling Options. *Materials* 2019, 12, 725.